# **APPENDICES**

## APPENDIX A

## List of chemical used

Acetone CH <sub>3</sub> COCH <sub>3</sub> Analar Buffer solution NaHCO <sub>3</sub> NH <sub>4</sub> HCO <sub>3</sub> Chloroform CHCl <sub>3</sub> Analar Cetytrimethlyammonium bromide(CTAB) [CH <sub>2</sub> (CH <sub>2</sub> ) <sub>15</sub> ](CH <sub>3</sub> ) <sub>3</sub> NBr Copper sulphate CuSO <sub>4.5</sub> H <sub>2</sub> O Decahydrophlene Dichloromethane CH <sub>2</sub> Cl <sub>2</sub> Analar Diethyl ether (C <sub>2</sub> H <sub>3</sub> ) <sub>2</sub> O Analar Disodium hydrogen phosphate Na <sub>2</sub> HPO <sub>4</sub> Ethanol CH <sub>3</sub> OH Analar Ethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3.</sub> H <sub>2</sub> O Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Potassium hydroxide KOH Potassium nitrate	Chemical name	Formula	Grade
Acetone CH <sub>3</sub> COCH <sub>3</sub> Analar Buffer solution NaHCO <sub>3</sub> NH <sub>4</sub> HCO <sub>3</sub> Chloroform CHCl <sub>3</sub> Analar Cetytrimethlyammonium bromide(CTAB) [CH <sub>2</sub> (CH <sub>2</sub> ) <sub>15</sub> ](CH <sub>3</sub> ) <sub>3</sub> NBr Copper sulphate CuSO <sub>4.5</sub> H <sub>2</sub> O Decahydrophlene Dichloromethane CH <sub>2</sub> Cl <sub>2</sub> Analar Diethyl ether (C <sub>2</sub> H <sub>3</sub> ) <sub>2</sub> O Analar Disodium hydrogen phosphate Na <sub>2</sub> HPO <sub>4</sub> Ethanol CH <sub>3</sub> OH Analar Ethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3.</sub> H <sub>2</sub> O Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Potassium hydroxide KOH Potassium nitrate			
Buffer solution NaHCO3 NH4HCO3 Chloroform CHCl3 Analar Cetytrimethlyammonium bromide(CTAB) [CH <sub>2</sub> (CH <sub>2</sub> ) <sub>15</sub> ](CH <sub>3</sub> ) <sub>3</sub> NBr Copper sulphate CuSO <sub>4.5</sub> H <sub>2</sub> O Decahydrophlene Dichloromethane CH <sub>2</sub> Cl <sub>2</sub> Analar Diethyl ether (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O Analar Disodium hydrogen phosphate Na <sub>2</sub> HPO <sub>4</sub> Ethanol CH <sub>3</sub> OH Analar Ethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3.</sub> H <sub>2</sub> O Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Potassium hydroxide KOH Potassium nitrate	Acetic acid (glacial)	CH₃COOH	Analar
NH4HCO3 Chloroform CHCl3 Chloroform	Acetone	CH <sub>3</sub> COCH <sub>3</sub>	Analar
Chloroform CHCl3 Analar Cetytrimethlyammonium bromide(CTAB) [CH <sub>3</sub> (CH <sub>2</sub> ) <sub>15</sub> ](CH <sub>3</sub> ) <sub>3</sub> NBr Copper sulphate CuSO <sub>4-5</sub> H <sub>2</sub> O  Decahydrophlene Dichloromethane CH <sub>2</sub> Cl <sub>2</sub> Analar Diethyl ether (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O Analar Disodium hydrogen phosphate Na <sub>2</sub> HPO <sub>4</sub> Ethanol CH <sub>3</sub> OH Analar Ethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3</sub> ,H <sub>2</sub> O  Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Potassium hydroxide KOH Potassium nitrate	Buffer solution .	NaHCO <sub>3</sub>	
Cetytrimethlyammonium bromide(CTAB) [CH <sub>3</sub> (CH <sub>2</sub> ) <sub>15</sub> ](CH <sub>3</sub> ) <sub>3</sub> NBr Copper sulphate CuSO <sub>4-5</sub> H <sub>2</sub> O  Decahydrophlene  Dichloromethane CH <sub>2</sub> Cl <sub>2</sub> Analar Diethyl ether (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O Analar Disodium hydrogen phosphate Na <sub>2</sub> HPO <sub>4</sub> Ethanol CH <sub>3</sub> OH Analar Diethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3</sub> ,H <sub>2</sub> O  Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH  2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol  Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether  Potassium hydroxide KOH  Potassium nitrate KNO <sub>4</sub>		NH <sub>4</sub> HCO <sub>3</sub>	
Copper sulphate         CuSO <sub>4-5</sub> H <sub>2</sub> O           Decahydrophlene         Dichloromethane         CH <sub>2</sub> Cl <sub>2</sub> Analar           Diethyl ether         (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O         Analar           Disodium hydrogen phosphate         Na <sub>2</sub> HPO <sub>4</sub> Ethanol           Ethanol         CH <sub>3</sub> OH         Analar           Ethylenediaminatetra-acetic acid (EDTA)         [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol         C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH           Ferric nitrate         Fe(NO <sub>3</sub> ) <sub>3</sub> .H <sub>2</sub> O         HCl         Analar           Methanol         CH <sub>2</sub> OH         2 - Methylpropanol - 2 - ol         (CH <sub>3</sub> ) <sub>3</sub> COH         Analar           n - butanol         Nitric acid         HNO <sub>3</sub> Analar           Oxalic acid dihydrate         C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O         Analar           Petrolium ether         Analar           Potassium hydroxide         KOH           Potassium nitrate         KNO <sub>4</sub>	Chloroform	CHCl <sub>3</sub>	Analar
Decahydrophlene Dichloromethane CH <sub>2</sub> Cl <sub>2</sub> Analar Diethyl ether (C <sub>2</sub> H <sub>3</sub> ) <sub>2</sub> O Analar Disodium hydrogen phosphate Ethanol Ethylenediaminatetra-acetic acid (EDTA) Ethylenediaminatetra-acetic acid (EDTA) CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3</sub> ,H <sub>2</sub> O Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Analar Potassium hydroxide KOH Potassium nitrate	Cetytrimethlyammonium bromide(CTAB)	$[CH_{3}(CH_{2})_{15}](CH_{3})_{3}NBr$	
Dichloromethane  CH <sub>2</sub> Cl <sub>2</sub> Analar  Diethyl ether  (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O Analar  Disodium hydrogen phosphate  Ethanol  Ethylenediaminatetra-acetic acid (EDTA)  Z - ethoxyethanol/ethylglycol  Ferric nitrate  Fe(NO <sub>3</sub> ) <sub>3</sub> ,H <sub>2</sub> O  Hydrochloric acid (12N)  Methanol  Z - Methylpropanol - 2 - ol  n - butanol  Nitric acid  Oxalic acid dihydrate  Petrolium ether  Potassium hydroxide  KOH  Potassium nitrate  CL <sub>2</sub> Cl <sub>2</sub> Analar  CL <sub>3</sub> OH  Analar  Analar  Analar  Analar  KOH	Copper sulphate	CuSO <sub>4.5</sub> H <sub>2</sub> O	
Diethyl ether (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O Analar Disodium hydrogen phosphate Na <sub>2</sub> HPO <sub>4</sub> Ethanol CH <sub>3</sub> OH Analar Ethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3</sub> ,H <sub>2</sub> O Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Potassium hydroxide KOH Potassium nitrate KNO <sub>4</sub>	Decahydrophlene		
Disedium hydrogen phosphate Na <sub>2</sub> HPO <sub>4</sub> Ethanol CH <sub>3</sub> OH Analar Ethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>3</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3</sub> ,H <sub>2</sub> O Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Potassium hydroxide KOH Potassium nitrate KNO <sub>4</sub>	Dichloromethane	CH <sub>2</sub> Cl <sub>2</sub>	Analar
Ethanol CH3OH Analar Ethylenediaminatetra-acetic acid (EDTA) [CH2N(CH2COOH)2]2 2 - ethoxyethanol/ethylglycol C2H3OCH2CH2OH Ferric nitrate Fe(NO3)3.H2O  Hydrochloric acid (12N) HCl Analar Methanol CH2OH 2 - Methylpropanol - 2 - ol (CH3)3COH Analar n - butanol  Nitric acid HNO3 Analar Oxalic acid dihydrate C2H2O4.2H2O Analar Petrolium ether Potassium hydroxide KOH Potassium nitrate KNO4	Diethyl ether	$(C_2H_5)_2O$	Analar
Ethylenediaminatetra-acetic acid (EDTA) [CH <sub>2</sub> N(CH <sub>2</sub> COOH) <sub>2</sub> ] <sub>2</sub> 2 - ethoxyethanol/ethylglycol C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH Ferric nitrate Fe(NO <sub>3</sub> ) <sub>3</sub> ,H <sub>2</sub> O  Hydrochloric acid (12N) HCl Analar Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol  Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar Petrolium ether Analar Potassium hydroxide KOH Potassium nitrate KNO <sub>4</sub>	Disodium hydrogen phosphate	$Na_2HPO_4$	
2 - ethoxyethanol/ethylglycol         C2H3OCH2CH2OH           Ferric nitrate         Fe(NO3)3,H2O           Hydrochloric acid (12N)         HCl         Analar           Methanol         CH2OH         2 - Methylpropanol - 2 - ol         (CH3)3COH         Analar           n - butanol         Nitric acid         HNO3         Analar           Oxalic acid dihydrate         C2H2O4-2H2O         Analar           Petrolium ether         Analar           Potassium hydroxide         KOH           Potassium nitrate         KNO4	Ethanol	CH <sub>3</sub> OH	Analar
Ferric nitrate         Fe(NO <sub>3</sub> ) <sub>3</sub> .H <sub>2</sub> O           Hydrochloric acid (12N)         HCI         Analar           Methanol         CH <sub>2</sub> OH         2 - Methylpropanol - 2 - ol         (CH <sub>3</sub> ) <sub>3</sub> COH         Analar           n - butanol         Nitric acid         HNO <sub>3</sub> Analar           Oxalic acid dihydrate         C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> .2H <sub>2</sub> O         Analar           Petrolium ether         Analar           Potassium hydroxide         KOH           Potassium nitrate         KNO <sub>4</sub>	Ethylenediaminatetra-acetic acid (EDTA)	$[\mathrm{CH_2N}(\mathrm{CH_2COOH})_2]_2$	
Hydrochloric acid (12N) HCl Analar  Methanol CH <sub>2</sub> OH  2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol  Nitric acid HNO <sub>3</sub> Analar  Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> -2H <sub>2</sub> O Analar  Petrolium ether Analar  Potassium hydroxide KOH  Potassium nitrate KNO <sub>4</sub>	2 - ethoxyethanol/ethylglycol	C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> OH	
Methanol CH <sub>2</sub> OH 2 - Methylpropanol - 2 - ol (CH <sub>3</sub> ) <sub>3</sub> COH Analar n - butanol Nitric acid HNO <sub>3</sub> Analar Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> .2H <sub>2</sub> O Analar Petrolium ether Analar Potassium hydroxide KOH Potassium nitrate KNO <sub>4</sub>	Ferric nitrate	$Fe(NO_3)_3.H_2O$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Hydrochloric acid (12N)	HCl	Analar
n - butanol  Nitric acid HNO3 Analar  Oxalic acid dihydrate C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> .2H <sub>2</sub> O Analar  Petrolium ether Analar  Potassium hydroxide KOH  Potassium nitrate KNO <sub>4</sub>	Methanol	CH <sub>2</sub> OH	
Nitric acid $HNO_3$ Analar Oxalic acid dihydrate $C_2H_2O_4.2H_2O$ Analar Petrolium ether Analar Potassium hydroxide $KOH$ Potassium nitrate $KNO_4$	2 - Methylpropanol - 2 - ol	(CH <sub>3</sub> ) <sub>3</sub> COH	Analar
Oxalic acid dihydrate $C_2H_2O_4.2H_2O$ Analar Petrolium ether Analar Potassium hydroxide KOH Potassium nitrate KNO4	n - butanol		
Petrolium ether Analar Potassium hydroxide KOH Potassium nitrate KNO <sub>4</sub>	Nitric acid	HNO <sub>3</sub>	Analar
Potassium hydroxide KOH Potassium nitrate KNO <sub>4</sub>	Oxalic acid dihydrate	$C_2H_2O_4.2H_2O$	Analar
Potassium nitrate KNO <sub>4</sub>	Petrolium ether		Analar
	Potassium hydroxide	КОН	
Potassium permanganate KMnO <sub>4</sub> Analar	Potassium nitrate	KNO <sub>4</sub>	
	Potassium permanganate	KMnO <sub>4</sub>	Analar

Chemical name	Formula	Grade
Potassium sulphate	K <sub>2</sub> SO <sub>4</sub>	
Reduction solution (Sodium sulphite)	N <sub>2</sub> SO <sub>4</sub> NaS	
Silver nitrate	AgNO <sub>4</sub>	
Sodium borate decahydrate	$Na_2B_4O-10H_2O$	
Sodium lauryl sulphate	CH <sub>3</sub> (H <sub>2</sub> ) <sub>4</sub> OSO <sub>2</sub> ONa	
Sulphuric acid (95% - 97%)	$H_2SO_4$	
10% vanilin in sulphuric acid (con.) .		

#### APPENDIX B

#### Preparation of solution

All preparation of solution are for 1 litre unless stated otherwise.

## B.1 Neutral detergent solution (NDS)

Ethylenediaminatetra - acetic acid (EDTA) (18.16 g) and 6.81 g of sodium borate decahydrate were placed together in a beaker. Approximately 200 ml distilled water was then added and the mixture heated in a water bath until dissolved (solution A). Sodium lauryl sulphate (30 g) was added to 10 ml 2 - ethoxyethanol/ethlglycol in another beaker (solution B). Then, the solution A were poured into solution B. Disodium hydrogen phosphate (4.56 g) were dissolved in 50 ml distilled water and then mixed together with solution A and B. The volume of the mixture was made up to about 50 ml and the pH adjusted to 6.9 - 7.1. Total volume of NDS was then made up to 1 litre in a volumetric flask.

## B.2 Acid detergent solution (ADS)

An ADS is made up of cetytrimethlyammonium bromide (CTAB) (20.0 g) in 1 L of 0.5 M sulphuric acid (27.7 ml in 1 L cons. sulphuric acid).

## B.3 Lignin reagents

## i - Saturated potassium permanganate

Potassium permanganate reagent (50.0 g) was dissolved in 1 L of distilled water. The solution was stored in a dark bottle at 4°C.

## ii - Lignin buffer solution

A mixture of ferric nitrate (6.0 g) and argentum nitrate (0.15 g) was dissolved in 100 ml distilled water (solution A). Glacial acetic acid (500 ml), potassium nitrate (5.0 g) and tertiary butyl alcohol (400 ml) was added sequantially to solution A and mixed thoroughly.

#### iii - Combined permanganate solution

Saturated potassium permanganate solution was mixed with lignin buffer solution (2:1; v/v) just prior to use and the mixture was stored in a dark bottle.

## iv - Demineralising solution

Oxalic acid dehydrate (5.0 g) was dissolved in 700 ml ethanol (95 %) before being mixed with 50 ml nitric acid (12 N) and 250 ml distilled water.

## APPENDIX C

# Location of deer farms in each state in Malaysia

STATE	LOCATION	SPECIES
Perlis	Titi tinggi (DVS)	Cervus timorensis
Kedah	Pendang (DVS).	Cervus timorensis
	Datai, Langkawi (P)	Cervus timorensis
	Anak Bukit (I)	Cervus timorensis
Perak	Grik (DVS)	Cervus timorensis
	Kg. Beng, Lenggong (S)	Cervus timorensis
	Taiping (Z)	Cervus timorensis
		Axis axis
	*	Axis kuhli
		Axis porcinus
		Cervus unicolor
	Ulu Kati (DVS)	Cervus timorensis
	Sg Siput (DVS)	Cervus timorensis
	Sungkai (WLNP)	Cervus unicolor
		Cervus timorensis
	Behrang Hulu (DVS)	Cervus timorensis
Kuala Lumpur	Taman Tasik Perdana (DBKL)	Dama dama
	Bang. Parlimen	Cervus timorensis
Negeri Sembilan	Serting (P)	Cervus timorensis
	Juasseh (P)	Cervus timorensis
	Siliau (DVS)	Cervus timorensis
	Gemenchih (P)	Cervus timorensis
	· · · · · · · · · · · · · · · · · · ·	

STATE	LOCATION	SPECIES
Selangor	Rawang (P)	Cervus timorensis
	Sg. Nilam (DVS)	Cervus timorensis
	Batu Arang (DVS)	Cervus timorensis
	Rasa (P)	Cervus timorensis
	Parit Baru (DVS)	Cervus timorensis
	Sg. Besar (DVS)	Cervus timorensis
	Kpg Jawa, Klang (I)	Cervus timorensis
	Shah Alam (I)	Cervus timorensis
	Ulu Kelang (Z)	Cervus timorensis
	Serdang (U)	Cervus timorensis
Melaka	Air Keroh (Z)	Axis kuhli
		Axis axis
		Cervus timorensis
		Cervus unicolor
Pahang	Bukit Redan (MARDI)	Cervus timorensis
	Rompin (P)	Cervus timorensis
	Taman Negara	Cervus unicolor
	Jenderak (WLNP)	Cervus timorensis
Johor	Ins. Haiwan Kluang (DVS)	Axis axis
		Cervus unicolor
	PORIM, Kluang	Cervus timorensis
	Segamat (P)	Cervus timorensis
	Desaru (SEDC)	Dama dama
	Taman Tunku (Z)	Cervus unicolor
	Johor Bahru	Cervus timorensis

STATE	LOCATION	SPECIES
Terengganu	Cermin Kiri (DVS)	Cervus timorensis
	Jerangau (DVS)	Cervus timorensis
		Axis axis
	Pusat Pertanian Sekayu	Cervus unicolor
		Cervus timorensis
Kelantan	Gua Musang (WLNP)	Cervus unicolor
		Cervus timorensis
	Jeram Pasu (DVS)	Cervus timorensis
Sarawak	Kuching (P)	Cervus unicolor
	Bintulu (DVS)	Cervus unicolor
Sabah	Keningau (DVS)	Axis axis
		Cervus unicolor
		Cervus timorensis
	Tawau (DVS)	Cervus unicolor

List of deer farm in Malaysia (Adapted from Vidyadaran et al.,1991)

DVS : Dept. of Veterinary Services

P : Private

I : Palace

Z : Zoological garden

DBKL : Dewan Bandaraya Kuala Lumpur

WLNP : Wildlife and National Park

MARDI : Malaysian Agriculture Research and Development Institute

SEDC : State Economy Development Corporation

#### APPENDIX D

# Artificial saliva for rumen simulating technique (RUSITEC)

3.92 % NaHCO<sub>3</sub>

9.30 % Na<sub>2</sub>HPO<sub>4</sub>.2H<sub>2</sub>O

9.40 % NaCl

11.4 % KCl

12.18 % MgCl<sub>2</sub>.6H<sub>2</sub>O

6.0 % CaCl<sub>2</sub>.2H<sub>2</sub>O

- \* NaHCO3 and Na2HPO4.2H2O were freshly prepared
- \* All chemical were added to make up 1L
- \* CaCl2.2H2O was added last
- \* The pH should be maintain around 9.5

APPENDIX E

# List of unidentified species of plants eaten by deer

Family	Genus	Vernicular Name
Euporbiaceae	Glochidion sp.	
Zingiberaceae	Zingiber sp.	
Lauraceae	Beilschmiedia sp.	
Myristicaceae	Horsefieldia sp.	
Gultifereae	Calophyllum sp.	
Menispermaceae		
Sterculiaceae		
Verbenaceae	Teijsmanniodendrons sp.	
Sarpotaceae	Palaqium rostratum	Nyatoh sidang
Dipterocarpaceae	Dipteracocarpus sp.	Keruing buluh
Symlpocaceae	Symplocos sp.	
Euphorbiaceae	Glochidion superbum	
Loganiaceae		
Oleaceae		
Celastraceae	Lophopetalum sp.	
Euhorbiaceae	Macaranga henni	
Euphorbiaceae	Macaranga cornifera	
Ulmaceae	Trema angustifolia	
Bombaceae	Durio sp.	
Verbenaceae	Callicarpa sp.	
Ulmaceae	Gironniera sp.	Hampas tebu
Euphorbiaceae	Macaranga sp.	
Euphorbiaceae	Aporusa sp.	
Ulmaceae	Trema sp.	

## APPENDIX F

# Reagents used in Metabolizable Energy (ME) determination

- \* Buffer solution (NaHCO3 and NH4HCO3)
- \* Macromineral solution
- \* Micromineral solution
- \* Reduction solution
- \* Resazurine solution

#### APPENDIX G

## Preparation of reagents for Thin Layer Chromatography (TLC)

## 1 - 10 % vanilin in concentrated H2SO4

Concentrated sulphuric acid and ethanol were mixed in the ration 2:1 (solution A). Vanilin powder (10 g) was then dissolved in 100 ml of solution A and stirred. The vanilin solution was prepared immediately before use.

# 2 - Dragendoff's reagent

Solution A: 1.7 g of bismuth nitrate in 100 ml water - acetic acid (80: 10)

Solution B: 4.0 g of potassium iodide in 70 ml distilled water

Solution C: 20 ml of acetic acid in 70 ml distilled water

\* Dragendoff's solution was made by adding 5 ml of Solution A and 5 ml of Solution B were added to Solution C before use.

#### APPENDIX H

# The analysis of varians (ANOVA) on PA for Sapium baccatum

Sample: Sapium baccatum

	Source	•		]	F-value					
	df	DM	CP	CF	EE	ASH	NFE	NDF	ADF	ADL
Model Error	3 8	190.20**	5.11*	580.73**	56.41	2.33 <sup>ns</sup>	85.32**	1.79 <sup>ns</sup>	8.22**	2.66 <sup>ns</sup>
Total	11									

	DM	CP	CF	EE	ASH	NFE	NDF	ADF	ADL
Mean :	31.58	21.18	18.13	4.71	5.22	51.10	27.81	17.00	8.20
S. E. :	2.12	0.73	3.17	0.51	0.14	2.97	1.07	1.19	0.35
C. V. :	23.27	11.90	60.52	37.67	9.13	20.12	13.39	24.21	14.88

## **ERRATA**

- For Table in page 15, 46, 47, 53, 55, 59, 60, 62, 64, 68, 72 and 96; the title for each
  Table should be on top of the table.
- 2. The title for Figure 5.1 (page 56) and Figure 5.3 (page 70) should be written below the Figure.