

N-(Pyrazin-2-yl)-4-toluidine

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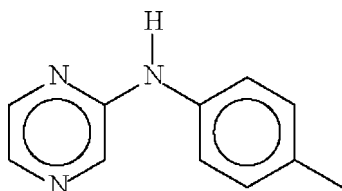
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.046; wR factor = 0.135; data-to-parameter ratio = 16.4.

The two aromatic systems in the title compound, $\text{C}_{11}\text{H}_{11}\text{N}_3$, are inclined by 19.1 (1)°, whilst the angle at the central amino N atom is 130.3 (2)°. The amino group forms a hydrogen bond to the pyrazine N-4 atom of an adjacent molecule, forming a chain motif.

Related literature

For the structure of aminopyrazine, see: Chao *et al.* (1976) and for that of *N*-(pyrazin-2-yl)-2-nitroaniline; see: Parsons *et al.* (2006). For two monoclinic modifications of *N*-(pyrazin-2-yl)-aniline, see: Abdullah & Ng (2008); Wan Saffiee *et al.* (2008).



Experimental

Crystal data

$\text{C}_{11}\text{H}_{11}\text{N}_3$	$b = 7.5323$ (3) Å
$M_r = 185.23$	$c = 12.0073$ (5) Å
Monoclinic, $C2/c$	$\beta = 105.790$ (3)°
$a = 21.7179$ (7) Å	$V = 1890.1$ (1) Å ³

$Z = 8$
Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹

$T = 100$ (2) K
 $0.30 \times 0.20 \times 0.05$ mm

Data collection

Bruker SMART APEX
diffractometer
Absorption correction: none
6057 measured reflections

2165 independent reflections
1437 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.041$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.046$
 $wR(F^2) = 0.135$
 $S = 1.03$
2165 reflections
132 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.27$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³

Table 1
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{N3}^i$	0.89 (2)	2.10 (2)	2.963 (2)	163 (2)

Symmetry code: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2008).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SG2277).

References

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supplementary materials

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N-(Pyrazin-2-yl)-4-toluidine

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Experimental

Chloropyrazine (1 ml, 1.1 mmol) and 4-toluidine (1.2 g, 1.1 mmol) were heated at 423–433 K for 3 h. The solid was dissolved in water. The compound was extracted with ether. The ether extract was dried over sodium sulfate; evaporation of the solvent gave colorless crystals among some unidentified dark brown materials.

Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95–0.98 Å) and were included in the refinement in the riding model approximation, with $U(\text{H})$ fixed at $1.2\text{--}1.5U(\text{C})$. The amino H-atom was located in a difference Fourier map, and was refined with a distance restraint of N—H 0.88 ± 0.01 Å.

Figures

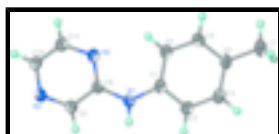


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) of $\text{C}_{11}\text{H}_{11}\text{N}_3$ at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

N-(Pyrazin-2-yl)-4-toluidine

Crystal data

$\text{C}_{11}\text{H}_{11}\text{N}_3$

$M_r = 185.23$

Monoclinic, $C2/c$

Hall symbol: $-C 2yc$

$a = 21.7179$ (7) Å

$b = 7.5323$ (3) Å

$c = 12.0073$ (5) Å

$\beta = 105.790$ (3)°

$V = 1890.1$ (1) Å³

$Z = 8$

$F_{000} = 784$

$D_x = 1.302$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 1368 reflections

$\theta = 2.9\text{--}27.2^\circ$

$\mu = 0.08$ mm⁻¹

$T = 100$ (2) K

Prism, colorless

$0.30 \times 0.20 \times 0.05$ mm

Data collection

Bruker SMART APEX
diffractometer

Radiation source: fine-focus sealed tube

1437 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.041$

supplementary materials

Monochromator: graphite $\theta_{\max} = 27.5^\circ$
 $T = 100(2)$ K $\theta_{\min} = 2.0^\circ$
 ω scans $h = -28 \rightarrow 28$
Absorption correction: None $k = -9 \rightarrow 9$
6057 measured reflections $l = -15 \rightarrow 15$
2165 independent reflections

Refinement

Refinement on F^2 Secondary atom site location: difference Fourier map
Least-squares matrix: full Hydrogen site location: inferred from neighbouring sites
 $R[F^2 > 2\sigma(F^2)] = 0.046$ H atoms treated by a mixture of independent and constrained refinement
 $wR(F^2) = 0.135$ $w = 1/[\sigma^2(F_o^2) + (0.0652P)^2 + 0.9346P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $S = 1.03$ $(\Delta/\sigma)_{\max} = 0.001$
2165 reflections $\Delta\rho_{\max} = 0.27 \text{ e } \text{\AA}^{-3}$
132 parameters $\Delta\rho_{\min} = -0.26 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods Extinction correction: none

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.31486 (7)	0.7286 (2)	0.14663 (14)	0.0223 (4)
H1	0.2746 (10)	0.754 (3)	0.1433 (17)	0.025 (5)*
N2	0.39338 (7)	0.5068 (2)	0.20972 (13)	0.0215 (4)
N3	0.30867 (7)	0.3328 (2)	0.31234 (14)	0.0240 (4)
C1	0.34916 (8)	0.8610 (2)	0.10751 (15)	0.0209 (4)
C2	0.31332 (8)	0.9994 (2)	0.04370 (15)	0.0225 (4)
H2	0.2680	0.9977	0.0274	0.027*
C3	0.34299 (9)	1.1385 (2)	0.00420 (15)	0.0243 (4)
H3	0.3176	1.2305	-0.0396	0.029*
C4	0.40936 (9)	1.1472 (2)	0.02707 (15)	0.0231 (4)
C5	0.44431 (8)	1.0095 (2)	0.09091 (15)	0.0226 (4)
H5	0.4897	1.0127	0.1081	0.027*
C6	0.41565 (8)	0.8672 (2)	0.13069 (15)	0.0222 (4)
H6	0.4412	0.7745	0.1734	0.027*
C7	0.44137 (10)	1.3010 (3)	-0.01539 (18)	0.0302 (5)
H7A	0.4876	1.2958	0.0203	0.045*
H7B	0.4244	1.4129	0.0056	0.045*
H7C	0.4328	1.2941	-0.0997	0.045*
C8	0.33543 (8)	0.5719 (2)	0.20318 (15)	0.0201 (4)
C9	0.29286 (8)	0.4818 (2)	0.25360 (16)	0.0231 (4)
H9	0.2514	0.5298	0.2450	0.028*
C10	0.36764 (8)	0.2678 (2)	0.32060 (16)	0.0231 (4)
H10	0.3813	0.1616	0.3628	0.028*

C11	0.40806 (9)	0.3538 (2)	0.26845 (16)	0.0235 (4)
H11	0.4487	0.3022	0.2742	0.028*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0217 (8)	0.0194 (8)	0.0284 (9)	0.0007 (6)	0.0110 (7)	0.0031 (7)
N2	0.0249 (8)	0.0191 (8)	0.0223 (8)	0.0013 (6)	0.0095 (6)	-0.0011 (6)
N3	0.0254 (8)	0.0213 (8)	0.0262 (9)	-0.0016 (6)	0.0087 (7)	0.0019 (7)
C1	0.0275 (9)	0.0177 (9)	0.0191 (9)	-0.0009 (7)	0.0089 (7)	-0.0013 (7)
C2	0.0240 (9)	0.0224 (10)	0.0214 (10)	0.0019 (8)	0.0067 (7)	-0.0018 (8)
C3	0.0344 (10)	0.0191 (9)	0.0194 (10)	0.0024 (8)	0.0072 (8)	0.0004 (7)
C4	0.0316 (10)	0.0193 (9)	0.0200 (9)	-0.0032 (8)	0.0096 (8)	-0.0008 (7)
C5	0.0252 (9)	0.0222 (10)	0.0221 (10)	-0.0018 (8)	0.0093 (8)	-0.0018 (8)
C6	0.0261 (9)	0.0207 (9)	0.0204 (9)	0.0004 (7)	0.0071 (7)	0.0004 (7)
C7	0.0390 (11)	0.0250 (11)	0.0287 (11)	-0.0033 (9)	0.0128 (9)	0.0045 (8)
C8	0.0237 (9)	0.0196 (9)	0.0177 (9)	-0.0013 (7)	0.0065 (7)	-0.0023 (7)
C9	0.0222 (9)	0.0211 (9)	0.0267 (10)	0.0003 (7)	0.0079 (8)	0.0002 (8)
C10	0.0275 (9)	0.0187 (9)	0.0237 (10)	0.0022 (7)	0.0077 (8)	0.0003 (8)
C11	0.0264 (9)	0.0211 (10)	0.0238 (10)	0.0020 (8)	0.0080 (8)	-0.0020 (8)

Geometric parameters (Å, °)

N1—C8	1.374 (2)	C4—C5	1.386 (3)
N1—C1	1.401 (2)	C4—C7	1.509 (3)
N1—H1	0.89 (2)	C5—C6	1.388 (2)
N2—C8	1.333 (2)	C5—H5	0.9500
N2—C11	1.344 (2)	C6—H6	0.9500
N3—C9	1.320 (2)	C7—H7A	0.9800
N3—C10	1.349 (2)	C7—H7B	0.9800
C1—C6	1.395 (2)	C7—H7C	0.9800
C1—C2	1.398 (2)	C8—C9	1.410 (2)
C2—C3	1.380 (3)	C9—H9	0.9500
C2—H2	0.9500	C10—C11	1.372 (3)
C3—C4	1.393 (3)	C10—H10	0.9500
C3—H3	0.9500	C11—H11	0.9500
C8—N1—C1	130.28 (15)	C5—C6—H6	120.2
C8—N1—H1	113.3 (13)	C1—C6—H6	120.2
C1—N1—H1	115.9 (14)	C4—C7—H7A	109.5
C8—N2—C11	115.60 (15)	C4—C7—H7B	109.5
C9—N3—C10	116.84 (15)	H7A—C7—H7B	109.5
C6—C1—C2	118.44 (16)	C4—C7—H7C	109.5
C6—C1—N1	124.95 (16)	H7A—C7—H7C	109.5
C2—C1—N1	116.58 (15)	H7B—C7—H7C	109.5
C3—C2—C1	120.81 (16)	N2—C8—N1	121.33 (16)
C3—C2—H2	119.6	N2—C8—C9	121.10 (16)
C1—C2—H2	119.6	N1—C8—C9	117.57 (15)
C2—C3—C4	121.50 (17)	N3—C9—C8	122.23 (16)

supplementary materials

C2—C3—H3	119.3	N3—C9—H9	118.9
C4—C3—H3	119.3	C8—C9—H9	118.9
C5—C4—C3	117.08 (16)	N3—C10—C11	120.51 (17)
C5—C4—C7	121.84 (16)	N3—C10—H10	119.7
C3—C4—C7	121.08 (17)	C11—C10—H10	119.7
C4—C5—C6	122.59 (16)	N2—C11—C10	123.68 (17)
C4—C5—H5	118.7	N2—C11—H11	118.2
C6—C5—H5	118.7	C10—C11—H11	118.2
C5—C6—C1	119.58 (17)		
C8—N1—C1—C6	-7.4 (3)	N1—C1—C6—C5	-177.39 (17)
C8—N1—C1—C2	174.81 (17)	C11—N2—C8—N1	179.29 (16)
C6—C1—C2—C3	0.2 (3)	C11—N2—C8—C9	-1.1 (2)
N1—C1—C2—C3	178.22 (16)	C1—N1—C8—N2	-13.4 (3)
C1—C2—C3—C4	-0.6 (3)	C1—N1—C8—C9	166.96 (17)
C2—C3—C4—C5	0.2 (3)	C10—N3—C9—C8	-1.1 (3)
C2—C3—C4—C7	-179.30 (17)	N2—C8—C9—N3	2.1 (3)
C3—C4—C5—C6	0.4 (3)	N1—C8—C9—N3	-178.20 (16)
C7—C4—C5—C6	179.96 (17)	C9—N3—C10—C11	-0.8 (3)
C4—C5—C6—C1	-0.7 (3)	C8—N2—C11—C10	-0.8 (3)
C2—C1—C6—C5	0.4 (3)	N3—C10—C11—N2	1.8 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots N3 ⁱ	0.89 (2)	2.10 (2)	2.963 (2)	163 (2)

Symmetry codes: (i) $-x+1/2, y+1/2, -z+1/2$.

Fig. 1

