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***N*-(Quinoxalin-2-yl)-4-toluidine**

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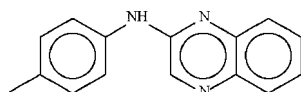
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å;
 R factor = 0.041; wR factor = 0.112; data-to-parameter ratio = 16.7.

The aromatic and the aromatic fused-rings in the title compound, $\text{C}_{15}\text{H}_{13}\text{N}_3$, open the angle at the planar N atom to 130.07 (13) and 129.98 (13) $^\circ$ in the two independent molecules in the asymmetric unit. The amino N atom of one molecule forms a hydrogen bond to the 4-N atom of an adjacent quinoxaliny ring, generating a supramolecular chain.

Related literature

For the structure of *N*-(2-pyridyl)-4-toluidine, see: Fairuz *et al.* (2008); for that of *N*-(pyrazin-2-yl)-4-toluidine, see: Wan Saffiee *et al.* (2008). The title compound is isostructural with *N*-(quinoxalin-2-yl)-4-chloroaniline; see: Idris *et al.* (2008).

**Experimental***Crystal data*

$\text{C}_{15}\text{H}_{13}\text{N}_3$	$V = 4872.5$ (6) Å ³
$M_r = 235.28$	$Z = 16$
Orthorhombic, <i>Pbca</i>	Mo $K\alpha$ radiation
$a = 12.2081$ (9) Å	$\mu = 0.08$ mm ⁻¹
$b = 11.3720$ (9) Å	$T = 100$ (2) K
$c = 35.097$ (3) Å	$0.40 \times 0.15 \times 0.05$ mm

Data collection

Bruker SMART APEX
diffractometer
Absorption correction: none
26747 measured reflections

5592 independent reflections
4089 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.051$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.112$
 $S = 1.03$
5592 reflections
335 parameters
2 restraints

H atoms treated by a mixture of
independent and constrained
refinement
 $\Delta\rho_{\text{max}} = 0.25$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.25$ e Å⁻³

Table 1Hydrogen-bond geometry (Å, $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{N5}$	0.89 (1)	2.26 (1)	3.114 (2)	163 (2)
$\text{N4}-\text{H4}\cdots\text{N2}^i$	0.87 (1)	2.19 (1)	3.017 (2)	157 (2)

Symmetry code: (i) $-x + \frac{3}{2}, y + \frac{1}{2}, z$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2009).

We thank the University of Malaya for supporting this study (FS358/2008 A).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK2339).

References

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supplementary materials

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N-(Quinoxalin-2-yl)-4-toluidine

W. A. M. Wan Saffiee, A. Idris, Z. Aiyub, Z. Abdullah and S. W. Ng

Comment

(type here to add)

Experimental

2-Chloroquinoxaline (1.64 g, 10 mmol) and 4-toluidine (1.07 g, 10 mmol) were mixed with ethanol (2 ml) and the mixture was heated at 423–433 K for 3 h. The product was dissolved in water and the solution extracted with ether. The ether phase was dried over sodium sulfate; the evaporation of the solvent gave well shaped crystals along with some unidentified brown material.

Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 0.98 Å) and were included in the refinement in the riding model approximation, with $U(\text{H})$ set to 1.2 to $1.5_{\text{eq}}U(\text{C})$. The amino H-atoms were located in a difference Fourier map, and were refined with a distance restraint of N—H 0.88 ± 0.01 Å; their temperature factors were freely refined.

Figures

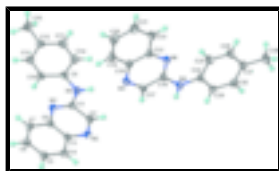


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) for the two independent molecules of $\text{C}_{15}\text{H}_{13}\text{N}_3$ at the 70% probability level. Hydrogen atoms are drawn as spheres of arbitrary radius.

N-(Quinoxalin-2-yl)-4-toluidine

Crystal data

$\text{C}_{15}\text{H}_{13}\text{N}_3$

$M_r = 235.28$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 12.2081$ (9) Å

$b = 11.3720$ (9) Å

$c = 35.097$ (3) Å

$V = 4872.5$ (6) Å³

$Z = 16$

$F_{000} = 1984$

$D_x = 1.283$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 3585 reflections

$\theta = 2.5$ – 27.6°

$\mu = 0.08$ mm⁻¹

$T = 100$ (2) K

Block, yellow

$0.40 \times 0.15 \times 0.05$ mm

supplementary materials

Data collection

Bruker SMART APEX diffractometer	4089 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.051$
Monochromator: graphite	$\theta_{\text{max}} = 27.5^\circ$
$T = 100(2)$ K	$\theta_{\text{min}} = 1.2^\circ$
ω scans	$h = -15 \rightarrow 15$
Absorption correction: None	$k = -14 \rightarrow 9$
26747 measured reflections	$l = -45 \rightarrow 45$
5592 independent reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.041$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.112$	$w = 1/[\sigma^2(F_o^2) + (0.0453P)^2 + 2.027P]$
$S = 1.03$	where $P = (F_o^2 + 2F_c^2)/3$
5592 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
335 parameters	$\Delta\rho_{\text{max}} = 0.25 \text{ e } \text{\AA}^{-3}$
2 restraints	$\Delta\rho_{\text{min}} = -0.25 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.48774 (11)	0.63997 (12)	0.55869 (4)	0.0200 (3)
N2	0.54428 (10)	0.38803 (11)	0.61730 (3)	0.0183 (3)
N3	0.41112 (10)	0.45078 (11)	0.55410 (3)	0.0169 (3)
N4	0.78134 (10)	0.95678 (12)	0.67401 (4)	0.0194 (3)
N5	0.57078 (11)	0.78948 (12)	0.62719 (4)	0.0207 (3)
N6	0.59133 (10)	0.96931 (11)	0.68290 (3)	0.0183 (3)
C1	0.47718 (12)	0.52591 (13)	0.57059 (4)	0.0166 (3)
C2	0.54419 (12)	0.49311 (14)	0.60265 (4)	0.0188 (3)
H2	0.5906	0.5510	0.6136	0.023*
C3	0.47637 (12)	0.30629 (13)	0.60042 (4)	0.0164 (3)
C4	0.47172 (12)	0.19136 (14)	0.61498 (4)	0.0193 (3)
H4A	0.5162	0.1697	0.6361	0.023*
C5	0.40277 (13)	0.11048 (14)	0.59870 (4)	0.0217 (3)
H5	0.3993	0.0328	0.6086	0.026*
C6	0.33726 (13)	0.14250 (14)	0.56743 (4)	0.0220 (3)
H6	0.2897	0.0859	0.5563	0.026*

C7	0.34082 (12)	0.25413 (14)	0.55266 (4)	0.0204 (3)
H7	0.2964	0.2741	0.5314	0.024*
C8	0.41034 (12)	0.33894 (13)	0.56900 (4)	0.0166 (3)
C9	0.43642 (12)	0.69945 (14)	0.52834 (4)	0.0178 (3)
C10	0.38087 (12)	0.64377 (14)	0.49874 (4)	0.0182 (3)
H10	0.3741	0.5606	0.4983	0.022*
C11	0.33549 (13)	0.71156 (14)	0.46985 (4)	0.0205 (3)
H11	0.2981	0.6731	0.4497	0.025*
C12	0.34273 (12)	0.83326 (14)	0.46935 (4)	0.0194 (3)
C13	0.39785 (13)	0.88706 (14)	0.49930 (5)	0.0222 (3)
H13	0.4036	0.9704	0.4999	0.027*
C14	0.44450 (13)	0.82189 (14)	0.52828 (5)	0.0222 (3)
H14	0.4823	0.8607	0.5483	0.027*
C15	0.29166 (14)	0.90520 (15)	0.43798 (5)	0.0269 (4)
H15A	0.3407	0.9702	0.4312	0.040*
H15B	0.2213	0.9369	0.4467	0.040*
H15C	0.2797	0.8553	0.4156	0.040*
C16	0.67680 (13)	0.92171 (13)	0.66615 (4)	0.0174 (3)
C17	0.66505 (13)	0.83080 (14)	0.63794 (4)	0.0198 (3)
H17	0.7294	0.7994	0.6266	0.024*
C18	0.47952 (13)	0.83502 (14)	0.64525 (4)	0.0190 (3)
C19	0.37461 (14)	0.79237 (15)	0.63617 (5)	0.0247 (4)
H19	0.3666	0.7318	0.6177	0.030*
C20	0.28394 (14)	0.83771 (16)	0.65375 (5)	0.0274 (4)
H20	0.2133	0.8084	0.6475	0.033*
C21	0.29513 (13)	0.92756 (16)	0.68096 (5)	0.0265 (4)
H21	0.2317	0.9591	0.6929	0.032*
C22	0.39656 (13)	0.97021 (15)	0.69051 (4)	0.0226 (4)
H22	0.4032	1.0306	0.7091	0.027*
C23	0.49085 (12)	0.92445 (14)	0.67284 (4)	0.0178 (3)
C24	0.81936 (13)	1.04794 (13)	0.69774 (4)	0.0177 (3)
C25	0.75245 (13)	1.11936 (15)	0.72024 (4)	0.0233 (4)
H25	0.6753	1.1083	0.7203	0.028*
C26	0.79967 (13)	1.20656 (15)	0.74242 (5)	0.0242 (4)
H26	0.7535	1.2550	0.7575	0.029*
C27	0.91210 (13)	1.22577 (14)	0.74339 (4)	0.0202 (3)
C28	0.97681 (12)	1.15459 (14)	0.72058 (4)	0.0191 (3)
H28	1.0539	1.1662	0.7204	0.023*
C29	0.93207 (12)	1.06744 (14)	0.69819 (4)	0.0180 (3)
H29	0.9786	1.0200	0.6829	0.022*
C30	0.96087 (14)	1.31961 (15)	0.76827 (5)	0.0263 (4)
H30A	1.0407	1.3193	0.7654	0.040*
H30B	0.9417	1.3042	0.7949	0.040*
H30C	0.9320	1.3965	0.7607	0.040*
H1	0.5235 (14)	0.6854 (15)	0.5750 (5)	0.038 (6)*
H4	0.8327 (12)	0.9193 (14)	0.6617 (5)	0.027 (5)*

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0240 (7)	0.0156 (7)	0.0204 (6)	-0.0025 (6)	-0.0051 (5)	0.0004 (5)
N2	0.0179 (6)	0.0195 (7)	0.0176 (6)	-0.0004 (5)	-0.0005 (5)	-0.0004 (5)
N3	0.0164 (6)	0.0162 (7)	0.0181 (6)	0.0003 (5)	0.0000 (5)	0.0003 (5)
N4	0.0166 (6)	0.0193 (7)	0.0223 (7)	-0.0015 (6)	0.0030 (5)	-0.0053 (5)
N5	0.0234 (7)	0.0180 (7)	0.0208 (6)	-0.0017 (6)	-0.0018 (5)	-0.0008 (5)
N6	0.0181 (6)	0.0189 (7)	0.0179 (6)	-0.0022 (5)	-0.0003 (5)	-0.0008 (5)
C1	0.0162 (7)	0.0163 (8)	0.0174 (7)	0.0000 (6)	0.0021 (6)	-0.0006 (6)
C2	0.0190 (7)	0.0195 (8)	0.0180 (7)	-0.0029 (6)	0.0000 (6)	-0.0020 (6)
C3	0.0147 (7)	0.0180 (8)	0.0164 (7)	0.0002 (6)	0.0024 (6)	-0.0010 (6)
C4	0.0198 (7)	0.0204 (8)	0.0177 (7)	0.0030 (6)	-0.0010 (6)	0.0023 (6)
C5	0.0244 (8)	0.0166 (8)	0.0240 (8)	0.0013 (7)	0.0004 (7)	0.0029 (7)
C6	0.0214 (8)	0.0187 (8)	0.0259 (8)	-0.0028 (7)	-0.0023 (6)	-0.0023 (7)
C7	0.0190 (8)	0.0210 (8)	0.0211 (7)	-0.0004 (7)	-0.0038 (6)	0.0001 (6)
C8	0.0152 (7)	0.0172 (8)	0.0174 (7)	0.0014 (6)	0.0015 (6)	0.0002 (6)
C9	0.0167 (7)	0.0184 (8)	0.0182 (7)	0.0003 (6)	0.0028 (6)	0.0011 (6)
C10	0.0205 (8)	0.0146 (8)	0.0194 (7)	0.0006 (6)	0.0032 (6)	-0.0017 (6)
C11	0.0219 (8)	0.0229 (9)	0.0167 (7)	0.0023 (7)	0.0009 (6)	-0.0016 (6)
C12	0.0183 (7)	0.0193 (8)	0.0206 (7)	0.0029 (6)	0.0045 (6)	0.0016 (6)
C13	0.0242 (8)	0.0151 (8)	0.0273 (8)	-0.0022 (6)	0.0021 (7)	0.0019 (7)
C14	0.0233 (8)	0.0197 (8)	0.0235 (8)	-0.0045 (7)	-0.0021 (6)	-0.0007 (7)
C15	0.0292 (9)	0.0241 (9)	0.0273 (9)	0.0034 (7)	-0.0026 (7)	0.0043 (7)
C16	0.0198 (7)	0.0155 (8)	0.0169 (7)	-0.0025 (6)	0.0003 (6)	0.0019 (6)
C17	0.0223 (8)	0.0177 (8)	0.0195 (7)	-0.0003 (7)	0.0012 (6)	-0.0019 (6)
C18	0.0225 (8)	0.0172 (8)	0.0175 (7)	-0.0023 (7)	-0.0026 (6)	0.0030 (6)
C19	0.0277 (9)	0.0235 (9)	0.0228 (8)	-0.0052 (7)	-0.0056 (7)	-0.0009 (7)
C20	0.0196 (8)	0.0334 (10)	0.0292 (9)	-0.0080 (7)	-0.0048 (7)	0.0012 (8)
C21	0.0199 (8)	0.0326 (10)	0.0270 (8)	-0.0012 (7)	0.0014 (7)	0.0003 (7)
C22	0.0218 (8)	0.0241 (9)	0.0219 (8)	-0.0012 (7)	0.0002 (6)	-0.0020 (7)
C23	0.0205 (8)	0.0171 (8)	0.0159 (7)	-0.0023 (6)	-0.0020 (6)	0.0027 (6)
C24	0.0211 (7)	0.0161 (8)	0.0159 (7)	-0.0018 (6)	-0.0009 (6)	0.0008 (6)
C25	0.0167 (7)	0.0251 (9)	0.0280 (8)	-0.0019 (7)	0.0019 (7)	-0.0059 (7)
C26	0.0220 (8)	0.0236 (9)	0.0269 (8)	-0.0008 (7)	0.0032 (7)	-0.0091 (7)
C27	0.0228 (8)	0.0184 (8)	0.0195 (7)	-0.0041 (6)	-0.0028 (6)	0.0009 (6)
C28	0.0169 (7)	0.0220 (8)	0.0184 (7)	-0.0028 (6)	-0.0018 (6)	0.0038 (6)
C29	0.0189 (7)	0.0183 (8)	0.0167 (7)	0.0029 (6)	0.0001 (6)	0.0011 (6)
C30	0.0252 (8)	0.0270 (9)	0.0269 (8)	-0.0060 (7)	-0.0014 (7)	-0.0060 (7)

Geometric parameters (\AA , $^\circ$)

N1—C1	1.369 (2)	C12—C15	1.507 (2)
N1—C9	1.4090 (19)	C13—C14	1.381 (2)
N1—H1	0.886 (9)	C13—H13	0.9500
N2—C2	1.301 (2)	C14—H14	0.9500
N2—C3	1.3793 (19)	C15—H15A	0.9800
N3—C1	1.3096 (19)	C15—H15B	0.9800

N3—C8	1.3751 (19)	C15—H15C	0.9800
N4—C16	1.365 (2)	C16—C17	1.439 (2)
N4—C24	1.4084 (19)	C17—H17	0.9500
N4—H4	0.872 (9)	C18—C19	1.406 (2)
N5—C17	1.299 (2)	C18—C23	1.411 (2)
N5—C18	1.382 (2)	C19—C20	1.368 (2)
N6—C16	1.314 (2)	C19—H19	0.9500
N6—C23	1.3745 (19)	C20—C21	1.405 (2)
C1—C2	1.440 (2)	C20—H20	0.9500
C2—H2	0.9500	C21—C22	1.371 (2)
C3—C4	1.405 (2)	C21—H21	0.9500
C3—C8	1.415 (2)	C22—C23	1.407 (2)
C4—C5	1.372 (2)	C22—H22	0.9500
C4—H4A	0.9500	C24—C29	1.394 (2)
C5—C6	1.406 (2)	C24—C25	1.397 (2)
C5—H5	0.9500	C25—C26	1.386 (2)
C6—C7	1.372 (2)	C25—H25	0.9500
C6—H6	0.9500	C26—C27	1.390 (2)
C7—C8	1.407 (2)	C26—H26	0.9500
C7—H7	0.9500	C27—C28	1.386 (2)
C9—C10	1.393 (2)	C27—C30	1.502 (2)
C9—C14	1.396 (2)	C28—C29	1.378 (2)
C10—C11	1.389 (2)	C28—H28	0.9500
C10—H10	0.9500	C29—H29	0.9500
C11—C12	1.387 (2)	C30—H30A	0.9800
C11—H11	0.9500	C30—H30B	0.9800
C12—C13	1.390 (2)	C30—H30C	0.9800
C1—N1—C9	130.07 (13)	C12—C15—H15B	109.5
C1—N1—H1	113.7 (13)	H15A—C15—H15B	109.5
C9—N1—H1	115.3 (13)	C12—C15—H15C	109.5
C2—N2—C3	116.66 (13)	H15A—C15—H15C	109.5
C1—N3—C8	116.08 (13)	H15B—C15—H15C	109.5
C16—N4—C24	129.98 (13)	N6—C16—N4	122.10 (14)
C16—N4—H4	115.4 (12)	N6—C16—C17	121.65 (14)
C24—N4—H4	114.5 (12)	N4—C16—C17	116.24 (14)
C17—N5—C18	116.44 (13)	N5—C17—C16	123.24 (14)
C16—N6—C23	116.15 (13)	N5—C17—H17	118.4
N3—C1—N1	122.78 (14)	C16—C17—H17	118.4
N3—C1—C2	121.74 (14)	N5—C18—C19	120.06 (14)
N1—C1—C2	115.48 (13)	N5—C18—C23	120.37 (14)
N2—C2—C1	123.17 (14)	C19—C18—C23	119.57 (15)
N2—C2—H2	118.4	C20—C19—C18	120.31 (15)
C1—C2—H2	118.4	C20—C19—H19	119.8
N2—C3—C4	119.67 (13)	C18—C19—H19	119.8
N2—C3—C8	120.02 (14)	C19—C20—C21	120.13 (15)
C4—C3—C8	120.31 (14)	C19—C20—H20	119.9
C5—C4—C3	119.80 (14)	C21—C20—H20	119.9
C5—C4—H4A	120.1	C22—C21—C20	120.72 (16)
C3—C4—H4A	120.1	C22—C21—H21	119.6

supplementary materials

C4—C5—C6	120.05 (15)	C20—C21—H21	119.6
C4—C5—H5	120.0	C21—C22—C23	120.01 (15)
C6—C5—H5	120.0	C21—C22—H22	120.0
C7—C6—C5	121.10 (15)	C23—C22—H22	120.0
C7—C6—H6	119.4	N6—C23—C22	118.66 (14)
C5—C6—H6	119.4	N6—C23—C18	122.09 (14)
C6—C7—C8	119.96 (14)	C22—C23—C18	119.25 (14)
C6—C7—H7	120.0	C29—C24—C25	118.58 (14)
C8—C7—H7	120.0	C29—C24—N4	116.69 (14)
N3—C8—C7	118.90 (13)	C25—C24—N4	124.72 (14)
N3—C8—C3	122.33 (14)	C26—C25—C24	119.36 (15)
C7—C8—C3	118.77 (14)	C26—C25—H25	120.3
C10—C9—C14	119.13 (14)	C24—C25—H25	120.3
C10—C9—N1	124.21 (14)	C25—C26—C27	122.46 (15)
C14—C9—N1	116.65 (14)	C25—C26—H26	118.8
C11—C10—C9	119.11 (14)	C27—C26—H26	118.8
C11—C10—H10	120.4	C28—C27—C26	117.18 (14)
C9—C10—H10	120.4	C28—C27—C30	121.66 (14)
C12—C11—C10	122.51 (15)	C26—C27—C30	121.15 (15)
C12—C11—H11	118.7	C29—C28—C27	121.58 (14)
C10—C11—H11	118.7	C29—C28—H28	119.2
C11—C12—C13	117.42 (14)	C27—C28—H28	119.2
C11—C12—C15	121.65 (15)	C28—C29—C24	120.82 (14)
C13—C12—C15	120.93 (15)	C28—C29—H29	119.6
C14—C13—C12	121.35 (15)	C24—C29—H29	119.6
C14—C13—H13	119.3	C27—C30—H30A	109.5
C12—C13—H13	119.3	C27—C30—H30B	109.5
C13—C14—C9	120.47 (15)	H30A—C30—H30B	109.5
C13—C14—H14	119.8	C27—C30—H30C	109.5
C9—C14—H14	119.8	H30A—C30—H30C	109.5
C12—C15—H15A	109.5	H30B—C30—H30C	109.5
C8—N3—C1—N1	179.27 (13)	C23—N6—C16—N4	-178.95 (14)
C8—N3—C1—C2	-0.5 (2)	C23—N6—C16—C17	2.1 (2)
C9—N1—C1—N3	-0.9 (2)	C24—N4—C16—N6	-4.0 (3)
C9—N1—C1—C2	178.94 (14)	C24—N4—C16—C17	174.97 (14)
C3—N2—C2—C1	0.4 (2)	C18—N5—C17—C16	-1.9 (2)
N3—C1—C2—N2	0.2 (2)	N6—C16—C17—N5	-0.2 (2)
N1—C1—C2—N2	-179.62 (14)	N4—C16—C17—N5	-179.18 (14)
C2—N2—C3—C4	-179.47 (14)	C17—N5—C18—C19	-177.95 (15)
C2—N2—C3—C8	-0.6 (2)	C17—N5—C18—C23	2.1 (2)
N2—C3—C4—C5	178.80 (14)	N5—C18—C19—C20	-179.52 (15)
C8—C3—C4—C5	-0.1 (2)	C23—C18—C19—C20	0.5 (2)
C3—C4—C5—C6	0.2 (2)	C18—C19—C20—C21	0.1 (3)
C4—C5—C6—C7	0.0 (2)	C19—C20—C21—C22	-0.6 (3)
C5—C6—C7—C8	-0.4 (2)	C20—C21—C22—C23	0.4 (3)
C1—N3—C8—C7	179.76 (13)	C16—N6—C23—C22	178.58 (14)
C1—N3—C8—C3	0.3 (2)	C16—N6—C23—C18	-1.9 (2)
C6—C7—C8—N3	-178.90 (14)	C21—C22—C23—N6	179.75 (15)
C6—C7—C8—C3	0.6 (2)	C21—C22—C23—C18	0.3 (2)

N2—C3—C8—N3	0.3 (2)	N5—C18—C23—N6	-0.2 (2)
C4—C3—C8—N3	179.13 (14)	C19—C18—C23—N6	179.87 (14)
N2—C3—C8—C7	-179.19 (13)	N5—C18—C23—C22	179.32 (14)
C4—C3—C8—C7	-0.3 (2)	C19—C18—C23—C22	-0.7 (2)
C1—N1—C9—C10	-15.1 (2)	C16—N4—C24—C29	-176.07 (15)
C1—N1—C9—C14	166.00 (15)	C16—N4—C24—C25	3.4 (3)
C14—C9—C10—C11	0.3 (2)	C29—C24—C25—C26	-0.4 (2)
N1—C9—C10—C11	-178.57 (14)	N4—C24—C25—C26	-179.90 (15)
C9—C10—C11—C12	-0.4 (2)	C24—C25—C26—C27	-0.3 (3)
C10—C11—C12—C13	-0.1 (2)	C25—C26—C27—C28	0.9 (2)
C10—C11—C12—C15	-179.21 (14)	C25—C26—C27—C30	-178.90 (15)
C11—C12—C13—C14	0.6 (2)	C26—C27—C28—C29	-0.8 (2)
C15—C12—C13—C14	179.69 (15)	C30—C27—C28—C29	179.05 (14)
C12—C13—C14—C9	-0.6 (2)	C27—C28—C29—C24	0.1 (2)
C10—C9—C14—C13	0.1 (2)	C25—C24—C29—C28	0.5 (2)
N1—C9—C14—C13	179.11 (14)	N4—C24—C29—C28	-179.91 (13)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...N5	0.89 (1)	2.26 (1)	3.114 (2)	163 (2)
N4—H4...N2 ⁱ	0.87 (1)	2.19 (1)	3.017 (2)	157 (2)

Symmetry codes: (i) $-x+3/2, y+1/2, z$.

Fig. 1

