ORIGINAL LITERARY WORK DECLARATION

Name of Candidate : SYAZWAN HANANI BINTI MERIAM SUHAIMY

(I.C/ Passport No :

Registration/ Matric No : KGG 100009

Name of Degree : MASTER OF ENGINEERING

Title of Project Paper/ Research Report/ Dissertation/ Thesis ("this work") :

SYNTHESIS AND CHARACTERIZATION OF AMORPHOUS CARBON NANOTUBES/ COPPER OXIDE HYBRID MATERIALS

Field of Study :

I do solemnly and sincerely declare that:

- (1) I am the sole author/writer of this Work;
- (2) This Work is original;
- (3) Any use of any work in which copyright exists was done by far of fair dealing and for permitted purposes and any excerpt or extract from, or reference to or reproduction of any copyright work has been disclosed expressly and sufficiently and the title of the Work and its authorship have been acknowledge in this Work;
- (4) I do not have any actual knowledge nor do I ought reasonably to know that the making of this work constitutes an infringement of any copyright work;
- (5) I hereby assign all and every rights in the copyright to this Work to the University of Malaya ("UM"), who henceforth shall be owner of the copyright

in this Work and that any reproduction or use in any form or by any means whatsoever is prohibited without the written consent of UM having been first had and obtained;

(6) I am fully aware that if in the course of making this Work I have infringed any Copyright whether intentionally or otherwise, I may be subject to legal action or any other action as may be determined by UM.

Candidate's Signature Date

Subscribed and solemnly declared before,

Witness's Signature Date

Name :

Designation:

ABSTRACT

Carbon nanotubes (CNTs) are designated as one of the most attractive material from researchers due to their unique properties and, wider range of applications. However, the synthesizing process of crystalline CNT is still a great issue because of their synthesizing process involving complicated process such as need higher operating temperature, catalyst support and expensive production cost. On the other hand, the synthesizing process of amorphous, carbon nanotubes (α -CNTs) offer a simple process. This will facilitate the producing of CNTs and open new doors to many potential applications. Therefore, we reported in this work a way of preparing α -CNT and hybridizing it with Cu₂O by a simple process using ferrocene and ammonium chloride at a low temperature (~250°C) in open atmosphere. The α -CNTs produced were purified with deionized for water and HCl for obtaining a high purity of CNTs. FESEM studies show that the morphology of the α -CNTs have changed after purification process. TEM studies revealed the diameter of the α -CNT and hybridized CNT. Raman studies revealed that the CNT is amorphous structure. EDX studies show the elements present in samples, and it is found that the major element is carbon. The X-ray diffraction pattern confirmed the amorphous nature of the sample and were also confirmed the presence of Cu₂O. The band gap of untreated, treated and hybridized samples were confirmed with UV-Vis studies.

ABSTRAK

Karbon nano tiub (CNTs) telah dikenalpasti sebagai salah satu bahan yang menarik dengan pelbagai sifat unik, dan sesuai untuk pelbagai aplikasi. Walau bagaimanapun, proses menghasilkan karbon nano tiub berstruktur hablur menjadi isu kerana proses penghasilannya adalah rumit seperti memerlukan suhu yang tinggi, memerlukan bantuan pemangkin, dan menelan kos yang tinggi. Walhal, proses penghasilan CNTs berstruktur amorfus (α -CNTs) hanya melalui proses yang ringkas. Perkara ini dapat memudahkan pemprosesan karbon nano tiub dan seterusnya membuka pintu baru kepada pelbagai aplikasi yang lebih baik. Oleh itu, kajian dijalankan untuk menghasilkan karbon nano tiub struktur jenis amorfus dan menghibridkannya dengan kuprum oksida (Cu₂O) melalui kaedah eksperimen yang ringkas menggunakan "ferrocene" dan ammonium klorida pada suhu yang rendah (~230°C). α-CNTs yang terhasil kemudiannya menjalani proses penulenan berasid untuk mendapatkan CNTs dengan ketulenan tinggi. Keputusan FE-SEM menunjukkan bahawa morfologi α -CNTs telah berubah setelah melalui proses penulenan. Diameter α -CNT telah diukur menggunakan TEM. Keputusan Raman telah mengesahkan bahawa karbon nano tiub yang dihasilkan adalah berstruktur amorfus. Analisis EDX telah menunjukkan jenis-jenis elemen yang terdapat di dalam sample dan didapati bahawa elemen utama adalah karbon. Kewujudan Cu₂O di dalam sampel telah dibuktikan melalui XRD, dan nilai "band gap" untuk semua sampel telah dikaji melalui eksperimen UV-Vis.

ACKNOWLEDGEMENT

With the deepest gratitude, I wish to thank every person who has came into my life and supported my journey and I honor every one of them.

Special thanks to my parents, Meriam Suhaimy A Rahman and Nor Asiah Mohammad for all their endless love and support. They are the precious jewels of my life, being my constant source of inspiration without which I would not have achieved this or any goal. Without their advice and support it would not have been possible for me to be in the position that I am currently in.

I would like to extend my gratitude to Dr. Mohd Rafie Johan for his generous professional advice, endless support and brilliant feedback when reviewing the research report to make sure I had it all right.

I am deeply thankful to all colleagues for continuous help in the project till completion. All of them are a tremendous of insight. I thank all for expanding my ability to understand more about this research. I also thank Puan Norzirah who has aided me during my research and for her support, sincerity and responsiveness to my needs.

TABLE OF CONTENTS

Title		Page
TITLE PAGE		i
ORIGINAL LITH	ERARY WORK DECLARATION	ii- iii
ABSTRACT		iv
ABSTRAK		v
ACKNOWLED	GEMENT	vi
TABLE OF CONTENTS		vii- viii
LIST OF FIGURES		ix-x
LIST OF TABLES		xi
LIST OF SYMBOLS AND ABBREVIATIONS		xii-xiii
CHAPTER ONI	E: INTRODUCTION	1
1.1	Background	1
1.2	Importance Research Problems	4
1.3	Research Objectives	5
1.4	Scope of Work	6
CHAPTER TW	O: LITERATURE REVIEW	2
2.1	Historical Developments	7-8
2.2	Type of Carbon Nanotubes	8-9
2.3	Structure and properties of Carbon Nanotubes	9-13
2.4	Synthesis Technique of Producing CNTs	13
	2.4.1 Methods for producing Crystalline CNTs	13
	2.4.1.1 Chemical Vapour Deposition(CVD)	14-15
	2.4.1 2 Arc Discharge	16-17

vii

	2.4.1 3 Laser Ablation	17-18
	2.4.1 4 Hydrothermal Synthesis	18-19
	2.4.2 Methods for producing Amorphous CNTs	20
	2.4.2.1 Chemical Vapour Deposition(CVD)	20
	2.4.2.2 Arc Discharge	20-21
	2.4.2.3 Template-Confines Growth	22-23
2.5	Hybrid Carbon Nanotubes	23-24
CHAPTER THE	REE: METHODOLOGY	25
3.1	Raw Materials	25
3.2	Preparation of Amorphous Carbon Nanotubes	26-28
3.3	Characterization Methods	29
	3.3.1 Morphological Studies	29
	3.3.2 Microstructural Studies	29
	3.3.3 Elemental Analysis	30
	3.3.4 Optical Studies	30-31
CHAPTER FOU	JR: RESULTS AND DISCUSSION	32
4.1	Morphological Studies	32-38
	4.1.1 FE-SEM analysis	32-35
	4.1.2 TEM analysis	36-38
4.2	Microstructural Studies	39-40
	4.2.1 XRD analysis	39-40
4.3	Elemental Studies	41-42
	4.3.1 EDX analysis	41-42
4.4	Optical Studies	43-50
	4.4.1 UV-Vis spectroscopy analysis	43-47
	4.4.2 Raman spectroscopy analysis	48-50
CHAPTER FIV	E: CONCLUSION	51-52
REFERENCES		53-57

LIST OF FIGURES

Figure 2.1:	Types of carbon nanotubes (a) SWCNTs; (b) MWCNTs and	
	(c) α-CNTs.	10
Figure 2.2:	Schematic diagram of typical CVD (Oncel, 2004).	15
Figure 2.3:	Schematic diagram of the arc discharge method(Meyappan, 2005)	16
Figure 3.1:	Flow chart of the work.	26
Figure 4.1:	FE-SEM images of the untreated α -CNTs with	33
	different magnifications: (a) 10,000X (b) 40,000X.	
Figure 4.2:	FE-SEM images of the treated α -CNTs at same magnification.	34
Figure 4.3:	FE-SEM images of the hybridized α -CNTs-Cu ₂ O sample at	35
	different weight of Cu ₂ O: (a) 0.2 g ; (b) 1.0 g .	
Figure 4.4:	TEM images of the treated α -CNTs at different magnifications:	36
	(a) Low magnification; (b) High magnification.	
Figure 4.5:	TEM images of the hybridized sample with 0.2 g Cu_2O at	37
	different magnifications: (a) Low magnification;	
	(b) High magnification.	
Figure 4.6:	TEM images of the hybridized sample with 1.0 g	38
	Cu ₂ O at different magnifications: (a) Low magnification;	
	(b) High magnification.	
Figure 4.7:	XRD pattern for: (a) Untreated α -CNTs; (b) Treated α -CNTs;	39
	(c) Hybridized sample with 0.2 g Cu_2O ;	
	(d) Hybridized sample with 1.0 g Cu_2O .	
Figure 4.8:	EDX spectra of the α -CNTs for (a) Untreated α -CNTs;	41
	(b) Treated α -CNTs.	
Figure 4.9:	UV-Vis absorption spectra for all samples at room temperature.	43
Figure 4.10:	Tauc/Davis-Mott plots for $(\alpha h \gamma)^{1/2}$ as a function of $h \gamma$ 45-	46
	for all samples: (a)Untreated sample; (b) Treated sample;	
	(c) Sample A; (d) Sample B; (e) Sample C; (f) Sample D.	

Figure 4.11:	UV-Vis transmittance spectra for all samples	47
	at room temperature.	
Figure 4.12:	Raman spectra for treated sample at room temperature.	48
Figure 4.13:	Raman spectra for hybrid sample with 0.2 g of Cu_2O	48
	at room temperature.	
Figure 4.14:	Raman spectra for hybrid sample with $1.0 \text{ g of } Cu_2O$	49
	at room temperature.	

LIST OF TABLES

Table 3.1:	Weight composition of Cu ₂ O and CNT.	28
Table 4.1:	Elemental analysis for untreated and treated samples.	42
Table 4.2:	Wavelength for all samples at room temperature.	44

LIST OF SYMBOLS AND ABBREVIATIONS

CNTs	Carbon nanotubes
α-CNTs	Amorphous carbon nanotubes
SWCNTs	Single-walled carbon nanotubes
DWCNTs	Double-walled carbon nanotubes
MWCNTs	Multi-walled carbon nanotubes
Cu ₂ O	Copper oxide
UV-Vis	Ultraviolet-visible
TEM	Transmission electron microscopy
FE-SEM	Field emission scanning electron microscopy
XRD	X-ray diffraction
EDX	Energy-dispersive X-ray
Eg	Bandgap energy
CVD	Chemical vapour deposition
PTFE	Polytetrafluoroethylene
AAO	Aluminium oxide templates
Co	Cobalt
Ni	Nickel
Fe	Iron
Не	Helium
FeS	Ferrous sulphide
DC	Direct current

PEG	Polyethylene glycol
ml	Millilitre
Ni-Al	Nickel aluminium alloy
h	Hour
min	Minute
$Fe(C_5H_5)_2$	Ferrocene
Nm	Nanometer
cm	Centimetre
G	Giga
Pa	Pascal
NH ₄ Cl	Ammonium chloride
HCl	Hydrochloric acid
C ₂ H ₅ OH	Ethanol
М	Molarity
kV	Kilovolts
Cu-K α	Copper K-alpha
Å	Ångström
А	Ampere
α	Absorption coefficient
hv	Photon energy of the incident light
n	Type of optical transition
В	Constant